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(54) Title: A PROCESS FOR MANUFACTURING POLY	YESTE	TP	

(57) Abstract

The present invention refers to a process for manufacturing polyester comprising at least the following steps: a) esterifying or transesterifying a carboxylic diacid or a diester of a carboxylic diacid with a diol, b) polymerizing in melt phase the esterification or transesterification product, c) solidifying and granulating, d) placing in contact the granules with a liquid swelling medium, e) post-condensing the granules in solid phase. The realization of the immersion step in a swelling medium permits increasing the kinetics of the solid phase post-condensation step.

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"A PROCESS FOR MANUFACTURING POLYESTER".

The present invention refers to the manufacturing of polyester-type polymers and refers particularly to the solid phase post-condensation steps.

- The manufacture of polyesters is generally carried out in several steps. The first step is a step of esterification or transesterification of a carboxylic diacid or a diacid diester with a diol. This step is followed by a melt phase polymerization step. As the
- reaction advances, the product becomes more and more viscous. Above a certain viscosity the technology used nowadays does not permit the continuation of the reaction. The viscosity limit attained in melt phase is generally around 85 ml/g (viscosity index). If
- 15 higher molecular weight compounds are to be obtained, it is necessary to proceed with the reaction in solid phase. This step is called a solid phase post-condensation step. It consists in proceeding with the condensation through heating in a gaseous medium or
- under vacuum, the polymer being in the form of solid elements. This step is very slow and constitutes a limiting step in the process of manufacturing a polyester.
- It is known that a limiting factor in solid phase post-condensation is the diffusion of condensation by-products within the polymeric mass. To speed up the solid phase post-condensation, US Patent No. 5,391,694 discloses a process which permits the increase of the exchange surface between the solid particles and the gaseous medium surrounding them by providing them with particular shapes during solidification. US Patent No.
- 4,755,587 teaches manufacturing porous tablets by compacting crystalline very light granulometry polymer powders. US Patent 5,532,335 teaches the improvement
- 35 in heat transfer within a solid phase post-

condensation medium by effecting the same in a liquid eventually pressurized medium and by increasing the post-condensation temperature.

Notwithstanding the progress reached, 5 always being made to obtain an improved performance solid phase post-condensation step. The object of the invention is to accelerate condensation kinetics by improving the diffusion condensation by-products outwards from within 10 particles. To attain this object, the invention employs a step preceding post-condensation, consisting in dipping the solidified polymer in a swelling medium.

Therefor, the invention proposes a process for 15 manufacturing polyester comprising at least the following steps:

- a) esterifying or transesterifying a carboxylic diacid or a diester of a carboxylic diacid with a diol,
- b) polymerizing in melt phase the esterification or20 transesterification product,
 - c) solidifying and granulating,
 - d) placing in contact the granules with a liquid swelling medium,
 - e) post-condensing the granules in solid phase.
- These steps constitute the essential operations necessary to carry out the process. The process may comprise others. Each of these steps may eventually be subdivided in elementary operations within several installations.
- Polymerization step b) is in fact widely applied by industry. It consists in a condensation in the molten state of the reaction product of step a). It is in general catalyzed by a metal compound, for example, antimonium trioxide. During the advancement of the
- 35 reaction, the product becomes more and more viscous.

diol.

Beyond a certain viscosity, the type of technology presently employed does not permit the reaction to proceed. The limit in viscosity attained in the molten phase is generally around 85 ml/g. If higher molecular weight compounds are desired, it is necessary to proceed with a solid phase condensation.

By viscosity it is to be understood the viscosity index (VI) in ml/g measured at 25°C with the help of a Ubbelohde type viscosimeter for a 0.005 g/ml solution

of polymer dissolved at 115°C in a mixture comprising 50% by weight of phenol and 50% by weight of 1,2-dichlorobenzene.

During step c) the molten polymer coming from step b) is made into the form of a granular solid, for example

- 15 by extrusion and cooling. It is then essentially amorphous. The granules exitting this step can be used either as they are, eventually after crystallization, or can be submitted to solid phase post-condensation for applications requiring higher molecular weight 20 polymers.
 - The solid phase post-condensation carried out during step e) consists in heating the granules coming from step c) to temperatures comprised between the glass transition temperature of the polymer and the melt temperature thereof. This condensation is accompanied by the separation of reaction products, especially of

In order to speed up the post-condensation kinetics, the present invention comprises a step d) of preparing the granules coming from step c). This step consists in placing in contact said granules with with a liquid swelling medium. The operation makes the granules become porous, thereby reducing the diffusional path, and thus increasing their specific exchange surface.

35 The diffusion of the diol coming from post

30

condensation is thus facilitated. Such a property permits speeding up the subsequent solid condensation step. The placing in contact may effected by any means, for example by immersion. The be effected either 5 step may continuously discontinuously.

As examples of liquid swelling media, the following compounds can be mentioned: methylene chloride, nitromethane, dioxane, acetone, benzene,

- dimethylformamide, dimentylacetamide, 10 methanol, ethanol, chloroform, trichloroethylene, tetrachloroethylene, carbon tetrachloride, toluene, benzyl alcohol, methyl-vinyl-ketone. Acetone particularly provides very good results.
- The above-described steps are necessary realization of the invention, but the process can include others, either upstream, downstream intermediately. The process according to the invention may particularly include a step of crystallizing the granules between the placing in contact and the solid phase condensation steps.

The swelling liquid medium may be used at temperature or maximum at a temperature lower than the boiling temperature of the liquid and the melt temperature of the treated polymer.

The placing in contact with the swelling medium may be having molecular weights applied to polymers advantageously higher than a molecular corresponding to a viscosity index around 30 ml/g. The process may be applied to polymers having higher viscosity indexes, especially until viscosity indexes around 90 ml/g. It may alternatively be applied to very high viscosity index polymers, for example up to 130 ml/g, in order to manufacture very high molecular 35 weight compounds.

The carrying out of the step of placing in contact with the swelling medium provides substantial technological improvements to the process. It can thus permit reducing the number of polymerization reactors.

- 5 It can permit simplifying or eliminating the crystallization procedures usually carried out. For example, since the classic process for preparing polymers after solidification and before post-condensation consists in a passage through three units
- 10 for pre-crystallization, crystallization and crystallization-conditioning the invention can reduce the number of operations so that not more than two placing in contact steps are employed, for example through immersion, and crystallization/conditioning.
- The process according to the present invention can be used to prepare polyester-type polymers. It can be applied to polyesters obtained from the following diacids: terephthalic acid, isophthalic acid, naphthalenedioic acid, and mixtures thereof, and the
- following diols: ethylene glycol, propylene glycol, butane diol, neopentyl glycol, diethylene glycol, bisphenol and mixtures thereof. It can be applied particularly to the manufacture of poy(ethylene terephthalate).
- 25 Other details and advantages of the invention will become apparent from the examples given hereunder for illustrative purposes only.

Example 1

Granules of poly(ethylene terephthalate) polymer

30 manufactured in a conventional manner, with an initial viscosity index (VI) of 73 ml/g, are dipped during 4 days in acetone, at a temperature of 20°C. They are dried at room temperature during 12 hours. They are then subjected to a solid phase post-condensation at

35 190°C during 20 hours under vacuum. The viscosity

index (final VI) of the obtained granules is measured. Example 2

Granules of poly(ethylene terephthalate) polymer manufactured in a conventional manner, with an initial viscosity index (VI) of 74,7 ml/g, are dipped during 2 hours in acetone, at a temperature of 50°C. They are dried at room temperature during 12 hours. They are then subjected to a solid phase post-condensation at 214°C during 31 hours under vacuum. The viscosity index (final VI) of the obtained granules is measured according to the procedures previously set forth.

Comparative examples

Comparative samples are subjected to the same heat treatment without immersion in acetone. The sample of comparative example 1 is submitted to a 20-hour post-condensation at 190°C. The sample of comparative example 2 is submitted to a 31-hour post-condensation at 214°C.

The results are respectively presented in tables 1 and 20 2. The efficiency of the solid phase post-condensation can be defined by the relationship between the differences in viscosity obtained during the post-condensation applied during the same period of time both for the compound having been submitted to an immersion and to a compound not having been submitted to immersion. Hence, for a compound not having been submitted to immersion (comparative examples) the efficiency is the reference efficiency, equal to 1.

Table 1

	Initial VI (ml/g)	Final VI (ml/g)	Difference	Efficiency
Comparative Example	73	88.3	15.3	1
Example 1	73	95.2	22.2	1.45

It appears that the invention permits attaining 45% of the solid phase post-condensation.

5 Table 2

	Initial VI (ml/g)	Final VI (ml/g)	Difference	Efficiency
Comparative Example	74.7	100.5	25.8	1
Example 2	734.7	108.9	34.2	1.33

It appears that the invention permits attaining 33% of the solid phase post-condensation.

CLAIMS

- 1 A process for manufacturing polyester comprising the following steps:
- a) esterifying or transesterifying a carboxylic diacid
 5 or a diester of a carboxylic diacid with a diol,
 - b) polymerizing in melt phase the esterification or transesterification product,
 - c) solidifying and granulating,
- d) placing in contact the granules with a liquid 10 swelling medium,
 - e) post-condensing the granules in solid phase.
 - 2 The process of claim 1, characterized by comprising a crystallization step between the steps of placing in contact with the liquid swelling medium and
- 15 the solid phase post-condensation.
 - 3 The process of one of claims 1 or 2, characterized in that the liquid swelling medium is selected from methylene chloride, dioxane, nitromethane, acetone, benzene, dimethylformamide, dimethylacetamide,
- 20 methanol, ethanol, chloroform, trichloroethylene, tetrachloroethylene, carbon tetrachloride, toluene, benzyl alcohol, methyl-vinyl-ketone.
 - 4 The process of claim 3, characterized in that the liquid swelling medium is acetone.
- 5 The process of any of claims 1 to 4, characterized in that the swelling medium is heated to a temperature under the melting point of the polymer.
 - 6 The process of any of claims 1 to 5, characterized in that the viscosity index of the compound resulting
- 30 from the polymerization step is higher than 30 ml/g.
 - 7 The process of any of claims 1 to 6, characterized in that the carboxylic diacids are selected from terephthalic acid, isophthalic acid, naphthalenedioic acid and mixtures thereof.
- 35 8 The process of any of claims 1 to 7, characterized

in that the diol is selected from ethylene glycol, propylene glycol, butane diol, neopentyl glycol, diethylene glycol, bisphenol and mixtures thereof.

9 - The process of any of claims 1 to 8, characterized in that the polyester is poly(ethylene terephthalate).

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A. CLASSIFICATION OF SUBJECT MATTER IPC 7 C08G63/78 C080 C08G63/80 C08G63/88 According to International Patent Classification (IPC) or to both national classification and IPC **B. FIELDS SEARCHED** Minimum documentation searched (classification system followed by classification symbols) IPC 7 C08G Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched Electronic data base consulted during the international search (name of data base and, where practical, search terms used) C. DOCUMENTS CONSIDERED TO BE RELEVANT Category ^e Citation of document, with indication, where appropriate, of the relevant passages Relevant to claim No. X US 3 390 134 A (KIBLER, CHARLES JACOB) 1-4,6-8 25 June 1968 (1968-06-25) claims 1-10 column 2, line 24 -column 3, line 49 X CHEMICAL ABSTRACTS, vol. 89, no. 12, 1-4,6-8 18 September 1978 (1978-09-18) Columbus, Ohio, US; abstract no. 90449, XP002112011 abstract & JP 53 054295 A (TEIJIN LTD.) 17 May 1978 (1978-05-17) US 5 164 478 A (LEE, CHERYLYN ET AL) 1-8 17 November 1992 (1992-11-17) claims 1-5; example column 2, line 36 -column 3, line 54 Further documents are listed in the continuation of box C. Patent family members are listed in annex. Special categories of cited documents: T' later document published after the international filing date on priority date and not in conflict with the application but cited to understand the principle or theory underlying the "A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier document but published on or after the International "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone filing date document which may throw doubte on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art. document referring to an oral disclosure, use, exhibition or document published prior to the international filing date but later than the priority date claimed "&" document member of the same patent family Date of the actual completion of the international search Date of mailing of the international search report 22 February 2000 28/02/2000 Name and mailing address of the ISA Authorized officer European Patent Office, P.B. 5818 Patentiaan 2 NL - 2280 HV Rijewijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Fax: (+31-70) 340-3016 Krische, D

Inten nal Application No PCT/BR 99/00099

2.6		PCT/BR 99/00099
	ation) DOCUMENTS CONSIDERED TO BE RELEVANT	
Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 5 744 578 A (DUH, BEN) 28 April 1998 (1998-04-28) claim 14; examples 1,3 column 3, line 60 -column 5, line 56	1,2,5-8
X	GB 1 101 900 A (THE GOODYEAR TIRE & RUBBER COMP.) 31 January 1968 (1968-01-31) the whole document	1,3,4, 6-9
X _.	DATABASE WPI Section Ch, Week 8011 Derwent Publications Ltd., London, GB; Class A23, AN 80-19011C XP002112015 & JP 55 013715 A (NIPPON ESTER CO LTD), 30 January 1980 (1980-01-30) abstract	1,3,4,6
Y .	US 5 225 448 A (MAIER, THOMAS R. ET AL) 6 July 1993 (1993-07-06) claims 1-20; example 1 column 3, line 10 -column 4, line 47 column 6, line 67 -column 8, line 42	1-3,6-9
Y	CHEMICAL ABSTRACTS, vol. 97, no. 6, 9 August 1982 (1982-08-09) Columbus, Ohio, US; abstract no. 39561, XP002112012 abstract & AGEEV, E.P. ET AL.: VESTN. MOSK. UNIV., SER. 2: KHIM., vol. 23, no. 2, 1982, pages 75-87,	1-3,6-9
X ~	US 4 613 664 A (TATE, S. ET AL.) 23 September 1986 (1986-09-23) claims 1-5; example 1 column 2, line 25 -column 3, line 47	1,5-9
A	CHEMICAL ABSTRACTS, vol. 113, no. 2, 9 July 1990 (1990-07-09) Columbus. Ohio, US; abstract no. 7046, XP002112013 abstract & JP 02 038422 A (TOYOBO CO.) 7 February 1990 (1990-02-07)	1,3,4, 6-8
A ·	US 4 755 587 A (RINEHART, VERNE R.) 5 July 1988 (1988-07-05) cited in the application abstract column 3, line 31 -column 7, line 34 -/	1-9

inte onal Application No PCT/BR 99/00099

		PCT/BR 99/00099		
	ation) DOCUMENTS CONSIDERED TO BE RELEVANT			
Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to daim No.		
	DATABASE WPI Section Ch, Week 9132 Derwent Publications Ltd., London, GB; Class A23, AN 91-234125 XP002112016 & JP 03 152137 A (ASAHI CHEM IND CO LTD), 28 June 1991 (1991-06-28) abstract	1,7		
	CHEMICAL ABSTRACTS, vol. 91, no. 6, 6 August 1979 (1979-08-06) Columbus, Ohio, US; abstract no. 40779, XP002112014 abstract & CZERWINSKI, WOJCIECH ET AL.: PRZEGL. WLOK., vol. 33, no. 1, 1979, pages 7-9,	1,3,9		
	US 4 080 317 A (MORAWETZ, GOTTFRIED ET AL) 21 March 1978 (1978-03-21) abstract; example 1	1,2,6-9		
		·		

Form PCT/ISA/210 (continuation of second sheet) (July 1992)

information on patent family members

Intern sal Application No PCT/BR 99/00099

		 1					3K 99/00099
	tent document In search report		Publication date		atent family member(s)		Publication date
US	3390134	A	25-06-1968	NONE			
JP	53054295	A	17-05-1978	NONE			
US	5164478	A	17-11-1992	NONE			
US	5744578	A	28-04-1998	AU	53159		22-05-1998
				BR Cn	97125 12356		26-10-1999 17-11-1999
				WO	98188		07-05-1998
				EP	09371		25-08-1999
			. هـ بو مناه هـ هـ هـ هـ مناه بنور مناه	PL	3331		08-11-1999
GB	1101900	A	 -	DE	15706		22-01-1970
				FR US	14468 34532		10-10-1966 01-07-1969
					J4532	+v A	01-0/-1909
	55013715 	Α	30-01-1980	NONE			·
US	5225448	Α	06-07-1993	AU	6612		13-07-1995
				AU	50788		02-06-1994
				CA De	21045 693104		20-05-1994 12-06-1997
	•			DE	693104		23-10-1997
				EP	05982	99 A	25-05-1994
				JP	62069	94 A	26-07-1994
				MX 	93069	56 A	31-05-1994
US	4613664	A	23-09-1986	JP	18087		10-12-1993
				JP	50211		23-03-1993
				JP JP	612932 18004		24-12-1986 12-11-1993
			٠.	JP	501112		12-11-1993
				JP	620017		07-01-1987
				JP	17555		23-04-1993
				JP	40455		27-07-1992
				JP Ep	6115752 02078		17-07-1986 07-01-1987
JP	2038422	A	07-02-1990	NONE	 -		
US	4755587	A	05-07-1988	AU	60319		08-11-1990
				AU	137478		29-09-1988
	•			CN Ep	103059 028454		25-01-1989 28-09-1988
				JP	6328901		25-11-1988
				ΜX	16414		20-07-1992
				US	487632		24-10-1989
				ZA	880208	37 A	26-10-1988
JP	3152137	A	28-06-1991	NONE			
US	4080317	A	21-03-1978	AU	346437		19-04-1973
				BE	77370		31-01-1972
	•			CA DD	97289		12-08-1975 05-01-1973
	•			DE	9489 215224		27-04-1972
						/1	-: VT 4//6

information on patent family members

Inten nal Application No PCT/BR 99/00099

		Publication date		atent family member(s)	Publication date	
			GB NL ZA	1371505 A 7114553 A 7106801 A	23-10-1974 25-04-1972 26-07-1972	